

# Strong Kleinman-Forbidden Second Harmonic Generation in Chiral Sulfide: La<sub>4</sub>InSbS<sub>9</sub>

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Supporting Information

**ABSTRACT:** A new chiral sulfide family,  $Ln_4InSbS_9$  (Ln = La, Pr, Nd), with its own structure type in space group  $P4_12_12$  or its enantiomorph  $P4_32_12$  has been synthesized by solid-state reaction. Remarkably, the La member shows the strongest Kleinman-forbidden second harmonic generation to date, with an intensity 1.5 times that of commercial AgGaS<sub>2</sub> at a laser wavelength of 2.05  $\mu$ m, and exhibits type-I phase-matchable behavior. Density functional theory calculations and ab initio molecular dynamics simulations suggest that lattice vibrations may be responsible for the origin and magnitude of the strong SHG effect.

he prerequisite of a nonlinear optical (NLO) material is that it be crystallographically noncentrosymmetric (NCS). Out of the total 32 point groups, there are 20 NCS point groups, of which the chiral 422 and 622 ones should have null second harmonic generation (SHG) response under the restriction of Kleinman symmetry.<sup>1</sup> Nevertheless, as a result of dispersion, very small Kleinman-forbidden SHG coefficients are observed. For example,  $\alpha$ -TeO<sub>2</sub> with 422 symmetry shows  $d_{14}(1.328 \ \mu m) \approx 0.36 \ pm/V$  or  $d_{14}(1.064 \ \mu m) \approx 0.61 \ pm/V$ ,<sup>2a-d,h</sup> and (K<sub>3</sub>I)[SmB<sub>12</sub>(GaS<sub>4</sub>)<sub>3</sub>] with 622 symmetry exhibits a powder SHG intensity 0.3 times that of potassium dihydrogen phosphate (KDP) at 1.94  $\mu$ m.<sup>2e</sup> However, violation cases are also found in large-sized single crystals of  $\alpha$ -SiO<sub>2</sub> (32 point group) and LiIO<sub>3</sub> (6 point group), for which the unexpected  $d_{14}$  values of 2.6 × 10<sup>-3</sup> pm/V<sup>2f</sup> and 0.24 pm/V,<sup>2g</sup> respectively, have been observed. It has also been known that when the second harmonic frequency approaches that of the electronic exciton resonance  $(2h\nu \approx E_g)$ , the small Kleinman-forbidden coefficient can become large.<sup>2a,h-1</sup> For example,  $\alpha$ -TeO<sub>2</sub> shows a large Kleinman-forbidden  $d_{14}$  value of ~4.32 pm/V at 0.659  $\mu$ m, which is 10 times larger than that at 1.328  $\mu$ m, because the second harmonic wavelength (0.33  $\mu$ m  $\approx$  3.76 eV) is very close to its optical band gap (3.75 eV).<sup>2a,i,m</sup>

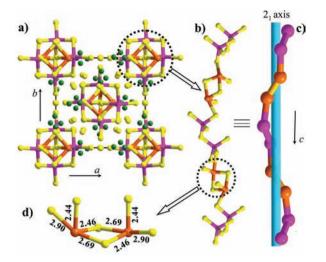
The common strategy to predesign a NCS compound is to involve asymmetric building units in the crystal structure, such as d<sup>0</sup> transition-metal cations (Ti<sup>4+</sup>, Nb<sup>5+</sup>, W<sup>6+</sup>, etc.) susceptible to second-order Jahn–Teller distortion or p cations with stereochemically active lone-pair electrons (Se<sup>4+</sup>, Te<sup>4+</sup>, As<sup>3+</sup>, Sb<sup>3+</sup>, etc).<sup>3</sup> We recently realized that in quaternary rare-earth metal/metal/antimony/chalcogenide systems, the SbQ<sub>2</sub> polyhedron tends to adopt an asymmetric geometry, such as the dimeric teeter-totter Sb<sub>2</sub>S<sub>6</sub> polyhedron in Ln<sub>4</sub>GaSbS<sub>9</sub>,<sup>4a</sup> the teeter-totter SbQ<sub>4</sub> (Q = S, Se) in La<sub>4</sub>FeSb<sub>2</sub>Q<sub>10</sub>,<sup>4b</sup> and squarepyramidal SbS<sub>5</sub> in Ln<sub>2</sub>Mn<sub>3</sub>Sb<sub>4</sub>S<sub>12</sub>.<sup>4c</sup> More interestingly, the arrangement of such asymmetric units is crucially affected by the MQ<sub>a</sub> polyhedron. For example, SbQ<sub>a</sub> units are eventually arranged in a centrosymmetric structure when linked by FeS<sub>4</sub> or MnS<sub>6</sub><sup>4b,c</sup> but an NCS structure with a strong SHG effect when linked by Ga<sub>2</sub>S<sub>7</sub>.<sup>4a</sup>

In this communication, we report an unusual chiral quaternary sulfide showing the strongest Kleinman-forbidden NLO response to date. Density functional theory (DFT) calculations and ab initio molecular dynamics (MD) simulations suggest that lattice vibrations may be responsible for its origin and magnitude.

The compounds Ln<sub>4</sub>InSbS<sub>9</sub> (Ln = La, Pr, Nd) crystallize in space group  $P4_12_12$  or its enantiomorph  $P4_32_12$  [Table S1 in the Supporting Information (SI)] and are characterized by unusual  $[In_2Sb_2S_{11}^{10-}]_{\infty}$  infinite helical chains propagating along the *c* direction and separated by isolated Ln<sup>3+</sup> cations and discrete S<sup>2-</sup> anions. These chains are further packed around the  $4_1$  helical axes (Figure 1a). Such chains are built from dimeric teeter-totter Sb<sub>2</sub>S<sub>6</sub> polyhedra and dual-apex-shared In<sub>2</sub>S<sub>7</sub> tetrahedra (Figure 1b); they are reminiscent of the chains in Ln<sub>4</sub>GaSbS<sub>9</sub><sup>4a</sup> but differ in that Sb<sub>2</sub>S<sub>6</sub> and In<sub>2</sub>S<sub>7</sub> are arranged around a 2-fold screw axis, which means that neighboring dimers are oriented in opposite directions (Figure 1c), whereas the neighboring Sb<sub>2</sub>S<sub>6</sub> or Ga<sub>2</sub>S<sub>7</sub> in Ln<sub>4</sub>GaSbS<sub>9</sub> are arranged in-phase (Figure S5 in the SI).

The SbS<sub>4</sub> polyhedron is remarkably distorted as a consequence of the stereochemically active lone-pair electrons of Sb<sup>3+</sup>, with Sb–S bond distances of 2.44–2.91 Å that are comparable to those found in Ln<sub>4</sub>GaSbS<sub>9</sub> (2.45–2.84 Å),<sup>4a</sup> La<sub>4</sub>FeSb<sub>2</sub>Q<sub>10</sub> (2.46–2.92 Å),<sup>4b</sup> and La<sub>7</sub>Sb<sub>9</sub>S<sub>24</sub> (2.43–3.00 Å).<sup>5</sup> The InS<sub>4</sub> tetrahedron is less distorted, with In–S distances of 2.45–2.49 Å and S–In–S angles ranging from 104 to 111°. These are comparable to those in CuInS<sub>2</sub>.<sup>6</sup> La<sup>3+</sup> cations and isolated S<sup>2–</sup> anions (sulfur atoms S7, S8, S9, and S10) occur between the infinite helical chains. Thus, the formula can be written as  $(La^{3+})_8([In_2Sb_2S_{11}]^{10-})(S^{2-})_7$ . The cationic La1 and La2 exhibit normal LaS<sub>6</sub> trigonal-prismatic coordination, and La3 and La4 are found in LaS<sub>7</sub> monocapped trigonal prisms with La–S lengths varying from 2.84 to 3.45 Å, comparable to those in  $\gamma$ -La<sub>2</sub>S<sub>3</sub>,<sup>7</sup>  $\alpha$ -La<sub>2</sub>S<sub>3</sub>,<sup>8</sup> La<sub>7</sub>Sb<sub>9</sub>S<sub>24</sub>,<sup>5</sup> and La<sub>4</sub>FeSb<sub>2</sub>S<sub>10</sub>.<sup>4b</sup>

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**Figure 1.** (a) View of La<sub>4</sub>InSbS<sub>9</sub> along (001). (b) Helical  $[In_2Sb_2S_{11}^{10-}]_{\infty}$  chains along the *c* direction. (c) Helical configuration of the chains. S atoms have been omitted for clarity, and the 2<sub>1</sub> screw axis is visualized as a thick blue line. (d) Local coordination of the dimeric teeter-totter Sb<sub>2</sub>S<sub>6</sub> polyhedra. Green, La; pink, In; orange, Sb; yellow, S.

Naturally, La<sub>4</sub>InSbS<sub>9</sub> is diamagnetic. Pr<sub>4</sub>InSbS<sub>9</sub> obeys the Curie–Weiss law over the entire experimental temperature region, whereas Nd<sub>4</sub>InSbS<sub>9</sub> deviates from it below 40 K (Figure S6). Their Curie constants and Weiss temperature are 6.54 and 7.25 emu K/mol and -21.67 and -22.74 K, respectively. The experimental effective magnetic moments are 7.23 and 7.61  $\mu_{\rm B}$ , respectively, which are comparable to the calculated values, 7.16 and 7.24  $\mu_{\rm B}$ .

The optical band gaps for the La, Pr, and Nd members were estimated to be 2.07, 2.09, and 2.12 eV, respectively, (Figure 2b

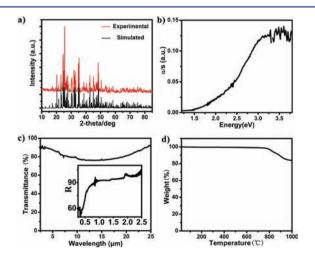


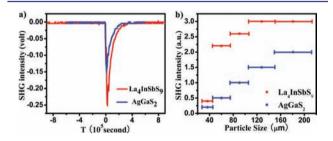
Figure 2. (a) Experimental and simulated X-ray diffraction patterns, (b) UV/vis/NIR diffuse reflectance spectrum, (c) IR and (inset) UV/ vis/NIR transmittance curves, and (d) thermogravimetric analysis curve for  $La_4InSbS_9$ .

and Figure S7), which are comparable to that of commercial  $AgGaS_2$  (2.62 eV),<sup>10</sup> implying that  $La_4InSbS_9$  may have a suitable laser damage threshold for NLO applications. Several Nd<sup>3+</sup> characteristic f–f transition absorptions between 1.3 and 2.1 eV were observed for Nd<sub>4</sub>InSbS<sub>9</sub>, similar to those for NaNdGa<sub>4</sub>Se<sub>8</sub>.<sup>11</sup> The diffuse-reflectance and IR spectroscopy

studies indicated that powdered La<sub>4</sub>InSbS<sub>9</sub> exhibits wide transparency comparable to that of powdered AgGaS<sub>2</sub> [1.0–25  $\mu$ m (Figure 2c) vs 0.6–25  $\mu$ m (Figure S8)]. These data distinguish La<sub>4</sub>InSbS<sub>9</sub> as a potential candidate for NLO materials in the mid- and far-IR regions.

The electronic structures of  $La_4InSbS_9$  were investigated using DFT (Figure S9). The valence band (VB) maximum and conduction band (CB) minimum are located at different kpoints indicating an indirect band gap of 2.23 eV, which is close to the experimental value. As the total and partial density of states (DOS) (Figure S10) indicate, the top of the VB is dominated by S 3p states, whereas the bottom of the CB is primarily derived from La 5d states mixed with some Sb 5p and In 5s states. Thus, the band gap absorption is likely to be the result of electronic transitions from S 3p to La 5d states.

Remarkably, La<sub>4</sub>InSbS<sub>9</sub> displays an SHG intensity 1.5 times that of commercial AgGaS<sub>2</sub> in an IR laser at 2.05  $\mu$ m (Figure



**Figure 3.** (a) Oscilloscope traces of the SHG signals of  $La_4InSbS_9$  and  $AgGaS_2$  at the same particle size of  $150-210 \ \mu m$ . (b) Particle-size dependence of the SHG intensity for  $La_4InSbS_9$  and  $AgGaS_2$ .

3). Moreover, the SHG intensities increase with the particle size and are saturated at a maximum value, indicating type-I phasematchable behavior. No obvious SHG signals were observed for the Pr and Nd members, possibly because of their much poorer crystallinity.

At first sight, this nonzero SHG response of La4InSbS9 is very surprising. As it crystallizes in space group P41212, which belongs to the 422 point group, the space-group symmetry requires two nonvanishing tensors of second-order susceptibilities to follow the equation  $d_{14} = -d_{25}$ . On the other hand, under the restriction of Kleinman symmetry,  $^{1}d_{14}$  must be equal to  $d_{25}$  (i.e.,  $d_{14} = d_{25}$ ). Consequently, both  $d_{14}$  and  $d_{25}$  must be equal to zero. Therefore, any material crystallizing in the 422 point group is forbidden from exhibiting an SHG response by Kleinman symmetry. Since Kleinman symmetry is based on the assumption that the medium is dispersionless, the presence of dispersion can damage its validity, leading to a very weak SHG response. For instance,  $\alpha$ -TeO<sub>2</sub> exhibits a small  $d_{14}$  of ~ 0.36 pm/V at 1.328  $\mu$ m.<sup>2a,h</sup> Electronic exciton resonance, which would be expected only when the second harmonic frequency is close to an absorption band gap of the material, can significantly enhance the SHG effect,<sup>12</sup> leading to strong violations of Kleinman symmetry.<sup>2a,h-1</sup> For instance,  $\alpha$ -TeO<sub>2</sub><sup>2a,i</sup> displays a large  $d_{14}$  value of 4.32 pm/V at 0.659  $\mu$ m, for which the second harmonic wavelength is 0.33  $\mu$ m (~3.76 eV), which is very close to its optical band gap (3.75 eV).<sup>2m</sup> However, this cannot explain the very large SHG coefficient of the title compound La4InSbS9 under our measurement conditions (roughly estimated to be 20 pm/V, which is 5 times that of  $\alpha$ -TeO<sub>2</sub>). In this case, the second harmonic wavelength is ~1.00  $\mu$ m (~1.2 eV), which differs significantly from the optical band gap (2.07 eV). Therefore, the electronic resonance

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would not be expected. As a further confirmation of this, we first calculated the second-order NLO susceptibility coefficients of La<sub>4</sub>InSbS<sub>9</sub> using the length-gauge formalism derived by Aversa and Sipe, and null static SHG coefficients were then obtained to satisfy the Kleinman relations in the zero-frequency limit<sup>13</sup> (for details, see the SI). Consequently, the second-order NLO coefficients of some possible configurations of La<sub>4</sub>InSbS<sub>9</sub> were calculated. These configurations were obtained via ab initio MD simulations at 300 K using the Nosé algorithm (for details, see the SI). The structure simulation length was 10 ps with a time step of 1 fs. After 7 ps, 11 typical configurations were chosen. The calculation results are summarized in Table S4. Nine of the 11 configurations had NLO coefficients close to 15 pm/V, which roughly agrees with the experimental observation: 1.5 times that of AgGaS<sub>2</sub> [i.e.,  $d_{36}(10.6 \ \mu m) =$ 13 pm/V).14 These results strongly suggest that thermal vibrations of the lattice induce structure configurational variations that may be responsible for the strong SHG effect of La<sub>4</sub>InSbS<sub>9</sub>.

Meanwhile, La<sub>4</sub>InSbS<sub>9</sub> exhibits excellent thermal stability and shows no obvious weight loss up to 765 °C (Figure 2d and Figure S11). The 16% weight loss above 765 °C corresponds to the decomposition and the volatilization of Sb<sub>2</sub>S<sub>3</sub> (mp 550 °C) and is close to the calculated value of 15.7% (for details, see the SI)

In summary, three quaternary chiral chalcogenides with their own structure type,  $Ln_4InSbS_9 \equiv (Ln^{3+})_8([In_2Sb_2S_{11}]^{10-})(S^{2-})_7$ (Ln = La, Pr, Nd) have been reported. The major structure motif is  $[In_2Sb_2S_{11}^{10-}]_{\infty}$  infinite helical chains propagating along the *c* direction that are well-separated by  $Ln^{3+}$  cations and discrete S<sup>2-</sup> anions. Remarkably, the La member shows the strongest Kleinman-forbidden SHG effects to date, with an intensity 1.5 times that of the commercial IR NLO material AgGaS<sub>2</sub> under type-I phase-matching conditions at a laser wavelength of 2.05  $\mu$ m. Meanwhile, the powdered La member exhibits a band gap of 2.07 eV, high transparency (1.00-25.00  $\mu$ m), and excellent thermal stability up to 765 °C. These primary data indicate that La<sub>4</sub>InSbS<sub>9</sub> is a potential candidate for IR NLO applications. The theoretical studies suggest that the origin and magnitude of the strong SHG response may originate from thermal vibrations of the lattice.

## ASSOCIATED CONTENT

#### **S** Supporting Information

Crystallographic data (CIF), experimental and theoretical methods, and additional tables and figures. This material is available free of charge via the Internet at http://pubs.acs.org.

# AUTHOR INFORMATION

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